

Dichloridobis(4-methylbenzyl)(1,10-phenanthroline- κ^2N,N')tin(IV)

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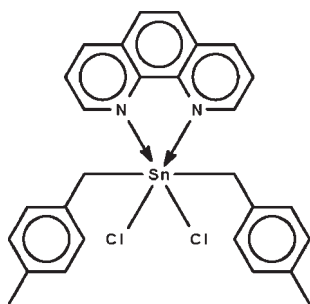
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 Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å;
 R factor = 0.025; wR factor = 0.074; data-to-parameter ratio = 19.6.

The tin(IV) atom in the title compound, $[\text{Sn}(\text{C}_8\text{H}_9)_2\text{Cl}_2(\text{C}_{12}\text{H}_8\text{N}_2)]$, is chelated by the N -heterocycle and bonded to two C atoms and two chloride ions in an $\text{SnC}_2\text{Cl}_2\text{N}_2$ octahedral coordination environment with the C atoms *trans* to each other. The dihedral angles between the heterocycle ring system and the benzene rings of the 4-methylbenzyl groups are 21.20 (12) and 28.71 (11)°.

Related literature

For the crystal structure of the 1,10-phenanthroline adduct with di(4-chlorobenzyl)tin dichloride, see: Tan *et al.* (2009). For the direct synthesis of substituted dibenzyltin dichlorides, see: Sisido *et al.* (1961).



Experimental

Crystal data

$[\text{Sn}(\text{C}_8\text{H}_9)_2\text{Cl}_2(\text{C}_{12}\text{H}_8\text{N}_2)]$
 $M_r = 580.10$
 Monoclinic, $P2_1/c$
 $a = 15.2180$ (2) Å
 $b = 10.4325$ (1) Å
 $c = 17.7990$ (2) Å
 $\beta = 115.0779$ (5)°

$V = 2559.42$ (5) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 1.23$ mm⁻¹
 $T = 293$ K
 $0.40 \times 0.30 \times 0.10$ mm

Data collection

Bruker SMART APEX CCD
 diffractometer
 Absorption correction: multi-scan
 (*SADABS*; Sheldrick, 1996)
 $T_{\min} = 0.640$, $T_{\max} = 0.887$

23764 measured reflections
 5882 independent reflections
 5066 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.023$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.025$
 $wR(F^2) = 0.074$
 $S = 1.05$
 5882 reflections

300 parameters
 H-atom parameters constrained
 $\Delta\rho_{\max} = 0.43$ e Å⁻³
 $\Delta\rho_{\min} = -0.68$ e Å⁻³

Table 1

Selected bond lengths (Å).

Sn1—C1	2.164 (2)	Sn1—N1	2.3780 (18)
Sn1—C9	2.179 (2)	Sn1—Cl1	2.5256 (6)
Sn1—N2	2.3595 (18)	Sn1—Cl2	2.5295 (7)

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *pubCIF* (Westrip, 2010).

We thank the University of Malaya (RG020/09AFR and PS338/2009C) for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5335).

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supplementary materials

Acta Cryst. (2010). E66, m304 [doi:10.1107/S1600536810005672]

Dichloridobis(4-methylbenzyl)(1,10-phenanthroline- κ^2N,N')tin(IV)

T. C. Keng, K. M. Lo and S. W. Ng

Experimental

Di(*p*-methylbenzyl)tin dichloride was synthesized by the reaction of *p*-methylbenzyl chloride and metallic tin (Sisido *et al.*, 1961). The reactant (0.45 g, 1 mmol) and 1,10-phenanthroline (0.18 g, 1 mmol) were heated in chloroform (50 ml) for 1 hour. Faint-yellow, almost colourless, blocks of (I) separated from the cool solution after a day.

Refinement

Hydrogen atoms were placed at calculated positions (C–H 0.93–0.96 Å) and were treated as riding on their parent atoms, with $U(H)$ set to 1.2–1.5 times $U_{eq}(C)$.

Figures

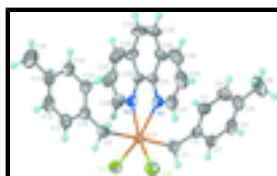


Fig. 1. The molecular structure of (I) at the 50% probability level. Hydrogen atoms are drawn as spheres of arbitrary radius.

Dichloridobis(4-methylbenzyl)(1,10-phenanthroline- κ^2N,N')tin(IV)

Crystal data

[Sn(C₈H₉)₂Cl₂(C₁₂H₈N₂)]

$M_r = 580.10$

Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

$a = 15.2180$ (2) Å

$b = 10.4325$ (1) Å

$c = 17.7990$ (2) Å

$\beta = 115.0779$ (5)°

$V = 2559.42$ (5) Å³

$Z = 4$

$F(000) = 1168$

$D_x = 1.505$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 9858 reflections

$\theta = 2.3$ – 28.3 °

$\mu = 1.23$ mm⁻¹

$T = 293$ K

Block, colourless

$0.40 \times 0.30 \times 0.10$ mm

Data collection

Bruker SMART APEX CCD
diffractometer

Radiation source: fine-focus sealed tube

5882 independent reflections

5066 reflections with $I > 2\sigma(I)$

supplementary materials

graphite $R_{\text{int}} = 0.023$
 ω scans $\theta_{\text{max}} = 27.5^\circ$, $\theta_{\text{min}} = 1.5^\circ$
Absorption correction: multi-scan
(SADABS; Sheldrick, 1996) $h = -19 \rightarrow 19$
 $T_{\text{min}} = 0.640$, $T_{\text{max}} = 0.887$ $k = -13 \rightarrow 13$
23764 measured reflections $l = -23 \rightarrow 22$

Refinement

Refinement on F^2 Primary atom site location: structure-invariant direct methods
Least-squares matrix: full Secondary atom site location: difference Fourier map
 $R[F^2 > 2\sigma(F^2)] = 0.025$ Hydrogen site location: inferred from neighbouring sites
 $wR(F^2) = 0.074$ H-atom parameters constrained
 $S = 1.05$ $w = 1/[\sigma^2(F_o^2) + (0.0413P)^2 + 0.9608P]$
5882 reflections where $P = (F_o^2 + 2F_c^2)/3$
300 parameters $(\Delta/\sigma)_{\text{max}} = 0.001$
0 restraints $\Delta\rho_{\text{max}} = 0.43 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.68 \text{ e } \text{\AA}^{-3}$

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Sn1	0.258495 (11)	0.590677 (12)	0.707814 (8)	0.03991 (6)
Cl1	0.28464 (5)	0.52229 (7)	0.85189 (4)	0.06375 (17)
Cl2	0.22061 (7)	0.82753 (6)	0.70201 (5)	0.0763 (2)
N1	0.24130 (14)	0.58938 (17)	0.56865 (11)	0.0451 (4)
N2	0.28083 (14)	0.38087 (17)	0.66953 (12)	0.0450 (4)
C1	0.41247 (18)	0.6286 (3)	0.75538 (14)	0.0521 (5)
H1A	0.4292	0.6956	0.7969	0.063*
H1B	0.4476	0.5519	0.7827	0.063*
C2	0.44531 (17)	0.6680 (2)	0.69126 (13)	0.0490 (5)
C3	0.4806 (2)	0.5793 (3)	0.65313 (16)	0.0568 (6)
H3	0.4881	0.4943	0.6706	0.068*
C4	0.5051 (2)	0.6148 (3)	0.58943 (18)	0.0673 (7)
H4	0.5296	0.5534	0.5656	0.081*
C5	0.49389 (19)	0.7397 (3)	0.56052 (16)	0.0674 (8)
C6	0.4612 (2)	0.8295 (3)	0.60007 (18)	0.0650 (7)
H6	0.4545	0.9145	0.5829	0.078*
C7	0.43823 (19)	0.7948 (2)	0.66475 (16)	0.0561 (6)
H7	0.4177	0.8573	0.6910	0.067*
C8	0.5137 (3)	0.7780 (5)	0.4875 (2)	0.1062 (15)
H8A	0.4542	0.8041	0.4425	0.159*
H8B	0.5405	0.7064	0.4705	0.159*
H8C	0.5590	0.8479	0.5031	0.159*
C9	0.10328 (18)	0.5559 (2)	0.66245 (16)	0.0528 (5)
H9A	0.0818	0.5933	0.7017	0.063*

H9B	0.0706	0.6016	0.6105	0.063*
C10	0.07041 (17)	0.4214 (2)	0.64869 (16)	0.0518 (5)
C11	0.0640 (2)	0.3462 (3)	0.71101 (17)	0.0671 (7)
H11	0.0804	0.3817	0.7632	0.081*
C12	0.0337 (2)	0.2200 (3)	0.6968 (2)	0.0732 (8)
H12	0.0306	0.1723	0.7398	0.088*
C13	0.0078 (2)	0.1629 (3)	0.6200 (2)	0.0702 (8)
C14	0.01504 (19)	0.2365 (3)	0.55829 (18)	0.0606 (6)
H14	-0.0008	0.2005	0.5063	0.073*
C15	0.04530 (17)	0.3623 (3)	0.57233 (15)	0.0532 (5)
H15	0.0490	0.4092	0.5293	0.064*
C16	-0.0291 (3)	0.0259 (4)	0.6033 (3)	0.1090 (13)
H16A	-0.0768	0.0130	0.6246	0.164*
H16B	0.0240	-0.0322	0.6301	0.164*
H16C	-0.0579	0.0103	0.5446	0.164*
C17	0.22739 (18)	0.6918 (3)	0.52146 (15)	0.0587 (6)
H17	0.2259	0.7720	0.5437	0.070*
C18	0.2148 (2)	0.6835 (4)	0.43852 (18)	0.0781 (9)
H18	0.2076	0.7575	0.4073	0.094*
C19	0.2132 (2)	0.5675 (4)	0.40508 (18)	0.0866 (11)
H19	0.2032	0.5611	0.3499	0.104*
C20	0.22645 (19)	0.4558 (4)	0.45285 (16)	0.0673 (8)
C21	0.24283 (16)	0.4721 (2)	0.53616 (13)	0.0470 (5)
C22	0.2244 (3)	0.3269 (5)	0.4227 (2)	0.0911 (12)
H22	0.2131	0.3146	0.3676	0.109*
C23	0.2380 (3)	0.2258 (4)	0.4710 (2)	0.0895 (12)
H23	0.2338	0.1444	0.4484	0.107*
C24	0.25889 (18)	0.2374 (3)	0.5567 (2)	0.0661 (8)
C25	0.26192 (15)	0.3617 (2)	0.58928 (15)	0.0466 (5)
C26	0.2746 (2)	0.1346 (3)	0.6110 (3)	0.0806 (10)
H26	0.2721	0.0513	0.5916	0.097*
C27	0.2933 (2)	0.1543 (2)	0.6910 (2)	0.0776 (9)
H27	0.3037	0.0854	0.7269	0.093*
C28	0.29680 (19)	0.2795 (2)	0.71911 (18)	0.0591 (6)
H28	0.3109	0.2929	0.7747	0.071*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Sn1	0.05738 (10)	0.03106 (8)	0.03553 (9)	-0.00481 (6)	0.02380 (7)	-0.00172 (5)
Cl1	0.0878 (4)	0.0687 (4)	0.0407 (3)	-0.0134 (3)	0.0330 (3)	0.0041 (3)
Cl2	0.1138 (6)	0.0339 (3)	0.1028 (6)	0.0027 (3)	0.0669 (5)	-0.0032 (3)
N1	0.0461 (10)	0.0525 (11)	0.0341 (9)	-0.0040 (8)	0.0145 (8)	0.0033 (7)
N2	0.0484 (10)	0.0345 (8)	0.0492 (10)	-0.0021 (7)	0.0178 (8)	-0.0019 (8)
C1	0.0595 (14)	0.0567 (13)	0.0350 (11)	-0.0159 (11)	0.0151 (10)	-0.0028 (10)
C2	0.0512 (12)	0.0536 (13)	0.0370 (10)	-0.0141 (10)	0.0138 (9)	-0.0036 (9)
C3	0.0604 (15)	0.0576 (15)	0.0492 (13)	-0.0031 (11)	0.0201 (11)	-0.0029 (10)
C4	0.0653 (16)	0.086 (2)	0.0521 (15)	-0.0013 (14)	0.0265 (13)	-0.0106 (14)

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C5	0.0528 (14)	0.104 (2)	0.0446 (13)	-0.0111 (15)	0.0199 (11)	0.0084 (14)
C6	0.0607 (15)	0.0698 (17)	0.0642 (16)	-0.0142 (13)	0.0261 (13)	0.0119 (14)
C7	0.0618 (14)	0.0512 (14)	0.0580 (14)	-0.0150 (11)	0.0282 (12)	-0.0063 (11)
C8	0.085 (2)	0.178 (5)	0.068 (2)	0.005 (3)	0.0453 (18)	0.036 (2)
C9	0.0527 (13)	0.0535 (13)	0.0542 (13)	0.0024 (11)	0.0247 (11)	0.0005 (11)
C10	0.0438 (12)	0.0594 (14)	0.0518 (13)	-0.0039 (10)	0.0198 (10)	0.0023 (10)
C11	0.0654 (16)	0.084 (2)	0.0537 (14)	-0.0161 (15)	0.0273 (12)	0.0028 (14)
C12	0.0649 (16)	0.083 (2)	0.0739 (19)	-0.0168 (15)	0.0317 (14)	0.0190 (16)
C13	0.0601 (16)	0.0616 (17)	0.089 (2)	-0.0098 (13)	0.0315 (15)	0.0026 (15)
C14	0.0525 (13)	0.0616 (15)	0.0671 (16)	-0.0088 (12)	0.0247 (12)	-0.0071 (13)
C15	0.0467 (12)	0.0590 (14)	0.0508 (13)	-0.0032 (11)	0.0179 (10)	0.0056 (11)
C16	0.129 (3)	0.064 (2)	0.150 (4)	-0.023 (2)	0.075 (3)	-0.001 (2)
C17	0.0593 (14)	0.0635 (15)	0.0491 (13)	-0.0048 (12)	0.0188 (11)	0.0174 (11)
C18	0.0656 (17)	0.115 (3)	0.0493 (15)	-0.0040 (17)	0.0203 (13)	0.0327 (17)
C19	0.073 (2)	0.151 (4)	0.0376 (14)	-0.010 (2)	0.0253 (13)	0.0004 (18)
C20	0.0550 (14)	0.110 (2)	0.0410 (13)	-0.0119 (15)	0.0244 (11)	-0.0197 (14)
C21	0.0416 (11)	0.0619 (14)	0.0388 (11)	-0.0079 (10)	0.0182 (9)	-0.0116 (10)
C22	0.082 (2)	0.134 (3)	0.065 (2)	-0.024 (2)	0.0394 (17)	-0.056 (2)
C23	0.081 (2)	0.095 (3)	0.100 (3)	-0.016 (2)	0.046 (2)	-0.062 (2)
C24	0.0499 (13)	0.0592 (16)	0.091 (2)	-0.0072 (12)	0.0309 (13)	-0.0364 (15)
C25	0.0384 (10)	0.0470 (12)	0.0549 (13)	-0.0060 (9)	0.0202 (9)	-0.0162 (10)
C26	0.0644 (17)	0.0374 (13)	0.133 (3)	0.0004 (12)	0.0348 (19)	-0.0237 (17)
C27	0.0646 (17)	0.0312 (12)	0.120 (3)	0.0017 (11)	0.0232 (18)	0.0057 (15)
C28	0.0574 (14)	0.0413 (12)	0.0707 (16)	0.0045 (11)	0.0195 (12)	0.0080 (11)

Geometric parameters (Å, °)

Sn1—C1	2.164 (2)	C11—C12	1.382 (4)
Sn1—C9	2.179 (2)	C11—H11	0.9300
Sn1—N2	2.3595 (18)	C12—C13	1.387 (4)
Sn1—N1	2.3780 (18)	C12—H12	0.9300
Sn1—C11	2.5256 (6)	C13—C14	1.383 (4)
Sn1—C12	2.5295 (7)	C13—C16	1.518 (5)
N1—C17	1.319 (3)	C14—C15	1.378 (4)
N1—C21	1.357 (3)	C14—H14	0.9300
N2—C28	1.332 (3)	C15—H15	0.9300
N2—C25	1.347 (3)	C16—H16A	0.9600
C1—C2	1.485 (3)	C16—H16B	0.9600
C1—H1A	0.9700	C16—H16C	0.9600
C1—H1B	0.9700	C17—C18	1.408 (4)
C2—C3	1.384 (4)	C17—H17	0.9300
C2—C7	1.393 (3)	C18—C19	1.345 (5)
C3—C4	1.386 (4)	C18—H18	0.9300
C3—H3	0.9300	C19—C20	1.406 (5)
C4—C5	1.384 (4)	C19—H19	0.9300
C4—H4	0.9300	C20—C21	1.406 (3)
C5—C6	1.385 (4)	C20—C22	1.443 (5)
C5—C8	1.507 (4)	C21—C25	1.440 (4)
C6—C7	1.385 (4)	C22—C23	1.320 (6)

C6—H6	0.9300	C22—H22	0.9300
C7—H7	0.9300	C23—C24	1.426 (5)
C8—H8A	0.9600	C23—H23	0.9300
C8—H8B	0.9600	C24—C26	1.394 (5)
C8—H8C	0.9600	C24—C25	1.413 (3)
C9—C10	1.475 (3)	C26—C27	1.345 (5)
C9—H9A	0.9700	C26—H26	0.9300
C9—H9B	0.9700	C27—C28	1.391 (4)
C10—C15	1.390 (4)	C27—H27	0.9300
C10—C11	1.395 (4)	C28—H28	0.9300
C1—Sn1—C9	178.47 (9)	C11—C10—C9	122.4 (2)
C1—Sn1—N2	91.02 (9)	C12—C11—C10	121.4 (3)
C9—Sn1—N2	90.27 (8)	C12—C11—H11	119.3
C1—Sn1—N1	91.80 (7)	C10—C11—H11	119.3
C9—Sn1—N1	89.41 (8)	C11—C12—C13	121.5 (3)
N2—Sn1—N1	70.10 (6)	C11—C12—H12	119.2
C1—Sn1—C11	88.62 (6)	C13—C12—H12	119.2
C9—Sn1—C11	90.50 (7)	C14—C13—C12	117.4 (3)
N2—Sn1—C11	92.93 (5)	C14—C13—C16	121.0 (3)
N1—Sn1—C11	163.03 (5)	C12—C13—C16	121.7 (3)
C1—Sn1—C12	91.59 (8)	C13—C14—C15	121.2 (3)
C9—Sn1—C12	87.42 (7)	C13—C14—H14	119.4
N2—Sn1—C12	162.30 (5)	C15—C14—H14	119.4
N1—Sn1—C12	92.32 (5)	C14—C15—C10	122.1 (2)
C11—Sn1—C12	104.63 (3)	C14—C15—H15	118.9
C17—N1—C21	119.1 (2)	C10—C15—H15	118.9
C17—N1—Sn1	125.14 (18)	C13—C16—H16A	109.5
C21—N1—Sn1	115.77 (14)	C13—C16—H16B	109.5
C28—N2—C25	118.8 (2)	H16A—C16—H16B	109.5
C28—N2—Sn1	123.61 (17)	C13—C16—H16C	109.5
C25—N2—Sn1	116.86 (15)	H16A—C16—H16C	109.5
C2—C1—Sn1	114.29 (15)	H16B—C16—H16C	109.5
C2—C1—H1A	108.7	N1—C17—C18	122.1 (3)
Sn1—C1—H1A	108.7	N1—C17—H17	118.9
C2—C1—H1B	108.7	C18—C17—H17	118.9
Sn1—C1—H1B	108.7	C19—C18—C17	119.2 (3)
H1A—C1—H1B	107.6	C19—C18—H18	120.4
C3—C2—C7	117.3 (2)	C17—C18—H18	120.4
C3—C2—C1	121.2 (2)	C18—C19—C20	120.5 (3)
C7—C2—C1	121.4 (2)	C18—C19—H19	119.7
C4—C3—C2	121.1 (3)	C20—C19—H19	119.7
C4—C3—H3	119.4	C21—C20—C19	116.9 (3)
C2—C3—H3	119.4	C21—C20—C22	118.1 (3)
C3—C4—C5	121.5 (3)	C19—C20—C22	125.0 (3)
C3—C4—H4	119.3	N1—C21—C20	122.1 (3)
C5—C4—H4	119.3	N1—C21—C25	118.45 (19)
C4—C5—C6	117.5 (2)	C20—C21—C25	119.4 (2)
C4—C5—C8	122.1 (3)	C23—C22—C20	122.0 (3)
C6—C5—C8	120.3 (3)	C23—C22—H22	119.0

supplementary materials

C7—C6—C5	121.1 (3)	C20—C22—H22	119.0
C7—C6—H6	119.4	C22—C23—C24	122.1 (3)
C5—C6—H6	119.4	C22—C23—H23	119.0
C6—C7—C2	121.3 (3)	C24—C23—H23	119.0
C6—C7—H7	119.4	C26—C24—C25	117.1 (3)
C2—C7—H7	119.4	C26—C24—C23	124.8 (3)
C5—C8—H8A	109.5	C25—C24—C23	118.1 (3)
C5—C8—H8B	109.5	N2—C25—C24	121.8 (2)
H8A—C8—H8B	109.5	N2—C25—C21	118.1 (2)
C5—C8—H8C	109.5	C24—C25—C21	120.2 (2)
H8A—C8—H8C	109.5	C27—C26—C24	120.8 (3)
H8B—C8—H8C	109.5	C27—C26—H26	119.6
C10—C9—Sn1	117.26 (17)	C24—C26—H26	119.6
C10—C9—H9A	108.0	C26—C27—C28	118.9 (3)
Sn1—C9—H9A	108.0	C26—C27—H27	120.5
C10—C9—H9B	108.0	C28—C27—H27	120.5
Sn1—C9—H9B	108.0	N2—C28—C27	122.6 (3)
H9A—C9—H9B	107.2	N2—C28—H28	118.7
C15—C10—C11	116.4 (2)	C27—C28—H28	118.7
C15—C10—C9	121.1 (2)		
C1—Sn1—N1—C17	-85.6 (2)	C11—C12—C13—C14	1.1 (5)
C9—Sn1—N1—C17	93.4 (2)	C11—C12—C13—C16	-177.6 (3)
N2—Sn1—N1—C17	-176.1 (2)	C12—C13—C14—C15	-1.0 (4)
Cl1—Sn1—N1—C17	-176.82 (15)	C16—C13—C14—C15	177.7 (3)
Cl2—Sn1—N1—C17	6.02 (19)	C13—C14—C15—C10	0.3 (4)
C1—Sn1—N1—C21	96.79 (17)	C11—C10—C15—C14	0.4 (4)
C9—Sn1—N1—C21	-84.15 (17)	C9—C10—C15—C14	179.6 (2)
N2—Sn1—N1—C21	6.36 (15)	C21—N1—C17—C18	-0.5 (4)
Cl1—Sn1—N1—C21	5.6 (3)	Sn1—N1—C17—C18	-177.98 (19)
Cl2—Sn1—N1—C21	-171.54 (15)	N1—C17—C18—C19	2.5 (4)
C1—Sn1—N2—C28	90.9 (2)	C17—C18—C19—C20	-1.6 (5)
C9—Sn1—N2—C28	-88.3 (2)	C18—C19—C20—C21	-1.1 (5)
N1—Sn1—N2—C28	-177.6 (2)	C18—C19—C20—C22	178.9 (3)
Cl1—Sn1—N2—C28	2.21 (19)	C17—N1—C21—C20	-2.4 (3)
Cl2—Sn1—N2—C28	-170.65 (16)	Sn1—N1—C21—C20	175.29 (18)
C1—Sn1—N2—C25	-99.24 (16)	C17—N1—C21—C25	177.6 (2)
C9—Sn1—N2—C25	81.58 (17)	Sn1—N1—C21—C25	-4.7 (3)
N1—Sn1—N2—C25	-7.69 (15)	C19—C20—C21—N1	3.2 (4)
Cl1—Sn1—N2—C25	172.09 (15)	C22—C20—C21—N1	-176.7 (2)
Cl2—Sn1—N2—C25	-0.8 (3)	C19—C20—C21—C25	-176.8 (2)
N2—Sn1—C1—C2	83.16 (19)	C22—C20—C21—C25	3.3 (4)
N1—Sn1—C1—C2	13.03 (19)	C21—C20—C22—C23	-0.5 (5)
Cl1—Sn1—C1—C2	176.07 (19)	C19—C20—C22—C23	179.6 (3)
Cl2—Sn1—C1—C2	-79.34 (19)	C20—C22—C23—C24	-2.2 (6)
Sn1—C1—C2—C3	-94.0 (3)	C22—C23—C24—C26	-179.6 (3)
Sn1—C1—C2—C7	83.4 (2)	C22—C23—C24—C25	2.0 (5)
C7—C2—C3—C4	-1.9 (4)	C28—N2—C25—C24	-0.2 (3)
C1—C2—C3—C4	175.6 (2)	Sn1—N2—C25—C24	-170.55 (17)
C2—C3—C4—C5	-1.0 (4)	C28—N2—C25—C21	178.7 (2)

C3—C4—C5—C6	2.8 (4)	Sn1—N2—C25—C21	8.3 (3)
C3—C4—C5—C8	-175.7 (3)	C26—C24—C25—N2	1.2 (4)
C4—C5—C6—C7	-1.6 (4)	C23—C24—C25—N2	179.7 (2)
C8—C5—C6—C7	176.9 (3)	C26—C24—C25—C21	-177.7 (2)
C5—C6—C7—C2	-1.3 (4)	C23—C24—C25—C21	0.9 (3)
C3—C2—C7—C6	3.0 (4)	N1—C21—C25—N2	-2.4 (3)
C1—C2—C7—C6	-174.5 (2)	C20—C21—C25—N2	177.6 (2)
N2—Sn1—C9—C10	16.46 (19)	N1—C21—C25—C24	176.5 (2)
N1—Sn1—C9—C10	86.56 (19)	C20—C21—C25—C24	-3.5 (3)
Cl1—Sn1—C9—C10	-76.48 (19)	C25—C24—C26—C27	-1.0 (4)
Cl2—Sn1—C9—C10	178.91 (19)	C23—C24—C26—C27	-179.5 (3)
Sn1—C9—C10—C15	-90.4 (3)	C24—C26—C27—C28	0.0 (5)
Sn1—C9—C10—C11	88.7 (3)	C25—N2—C28—C27	-1.0 (4)
C15—C10—C11—C12	-0.3 (4)	Sn1—N2—C28—C27	168.7 (2)
C9—C10—C11—C12	-179.5 (3)	C26—C27—C28—N2	1.1 (5)
C10—C11—C12—C13	-0.4 (5)		

Fig. 1

